

The image is a vertical rectangular greeting card. The background is a light gray color with a repeating pattern of small, light gray geometric shapes, specifically squares containing star-like patterns. In the upper half, there are several gold-colored stars of varying sizes hanging from thin, vertical gold lines. The lower half of the card features a large, stylized, light gray archway with a gold-colored outline. Inside this archway, the words "EID" and "MUBARAK" are written in a gold-colored, serif font, stacked vertically.

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Emulsions Tharwat F. Tadros, 2016-03-21 Chapter 1 General Introduction Definition of emulsions and the role of the emulsifier Classification based on the nature of the emulsifier Classification based on the structure of the system General instability problems with emulsions creaming sedimentation flocculation Ostwald ripening coalescence and phase inversion Importance of emulsions in various industrial applications Chapter 2 Thermodynamics of Emulsion Formation and Breakdown Application of the second law of thermodynamics for emulsion formation Balance of energy and entropy and non spontaneous formation of emulsions Breakdown of the emulsion by flocculation and coalescence in the absence of an emulsifier Role of the emulsifier in preventing flocculation and coalescence by creating an energy barrier resulting from the repulsive energies between the droplets Chapter 3 Interaction Forces between Emulsion Droplets Van der Waals attraction and its dependence on droplet size Hamaker constant and separation distance between the droplets Electrostatic repulsion resulting from the presence of electrical double layers and its dependence on surface or zeta potential and electrolyte concentration and valency Combination of the van der Waals attraction with double layer repulsion and the theory of colloid stability Steric repulsion resulting from the presence of adsorbed non ionic surfactants and polymers Combination of van der Waals attraction with steric repulsion and the theory of steric stabilisation Chapter 4 Adsorption of Surfactants at the Oil Water Interface Thermodynamic analysis of surfactant adsorption and the Gibbs adsorption isotherm Calculation of the amount of surfactant adsorption and area per surfactant molecule at the interface Experimental techniques for measuring the interfacial tension Chapter 5 Mechanism of Emulsification and the Role of the Emulsifier Description of the factors responsible for droplet deformation and its break up Role of surfactant in preventing coalescence during emulsification Definition of the Gibbs dilational elasticity and the Marangoni effect in preventing coalescence Chapter 6 Methods of Emulsification Pipe flow static mixers and high speed stirrers rotor stator mixer Laminar and turbulent flow Membrane emulsification High pressure homogenisers and ultrasonic methods Chapter 7 Selection of Emulsifiers The hydrophilic lipophilic balance HLB and its application in surfactant selection Calculation of HLB numbers and the effect of the nature of the oil phase The phase inversion temperature PIT method for emulsifier selection The cohesive energy ratio method for emulsifier selection Chapter 8 Creaming Sedimentation of Emulsions and its prevention Driving force for creaming sedimentation effect of gravity droplet size and density difference between the oil and continuous phase Calculation of the rate of creaming sedimentation in dilute emulsions Influence of increase of the volume fraction of the disperse phase on the rate of creaming sedimentation Reduction of creaming sedimentation Balance of the density of the two phases reduction of droplet size and effect of addition of thickeners Chapter 9 Flocculation of Emulsions and its Prevention Factors affecting flocculation Calculation of fast and slow flocculation rate Definition of stability ratio and its dependence on electrolyte concentration and valency Definition of the critical coagulation concentration and its dependence on electrolyte valency

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Rheology Essentials of Cosmetic and Food Emulsions Rüdiger Brummer,2006-05-27
Cosmetic emulsions exist today in many forms for a wide variety of applications including face and hand creams for normal dry or oily skin body milks and lotions as well as sun block products Keeping track of them and their properties is not always easy despite informative product names or partial names e g hand or face cream that clearly indicate their use and properties This practical manual provides a detailed overview that describes the key properties and explains how to measure them using modern techniques Written by an expert in flows and flow properties it focuses on the application of rheological flow measurements to cosmetic and food emulsions and the correlation of these results with findings from other tests Beginning with a brief history of rheology and some fundamental principles the manual describes in detail the use of modern viscometers and rheometers including concise explanations of the different available instruments But the focus remains on practical everyday lab procedures how to characterize cosmetic and food emulsions with different rheological tests such as temperature time stress and strain both static and dynamic Also the critical topic of how the results correlate with other important product characteristics for instance skin sensation pumping performance stability etc is carefully explored Many pictures illustrations graphs and tables help readers new to the measurement of cosmetic emulsions in their daily work as well as to the more experienced who seek additional special tips and tricks *Laws Relating to Interstate Commerce and Transportation* United States,1961 Fundamentals of Interface and Colloid Science J. Lyklema,2000-07-10 Fundamentals of Interface and Colloid Science FICS is a standard reference work with an educational nature The emphasis is on the basic facts and phenomena which are systematically explained FICS aims to make interface and colloid science accessible to a

wide audience Interface and colloid science is an important and fascinating field but one that is often overlooked and undervalued It has applications as diverse as agriculture mineral dressing oil recovery industrial chemistry medical science and biotechnology A deductive approach is followed with systems of growing complexity being treated as the book progresses Volume I Fundamentals 1st ed 1991 2nd ed 1993 reviews the physical chemistry required to understand current literature on interfacial and colloid science The volume starts from first principles and gradually increases the level Volume II Solid Liquid Interfaces 1995 treats the subject systematically for the first time including adsorption double layers and electronkinetics Volume III Interface Tension covers interfacial tensions monolayers and wetting Accessible to a wide audience without a detailed knowledge of physics and chemistry Complex mathematical derivations are kept to a minimum Treats interfacial and colloidal phenomena from first principles advanced command of physics and chemistry not required Takes the reader from elementary to expert level Acts as a reference and a textbook Contains extensive and detailed cumulative subject index

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